1/n-expansion for a hydrogen atom in pa
for a ground state, $B = 1$ ($F_{*} = 0.3449$)

	$F = 0.2$		$F = 0.24$			
	E_{I}	$\Gamma/2$	E_{I}	$\Gamma/2$		
	0.888	0	0.907	0	0.922	0
	0.898	0.031	0.885	0.055	0.883	0.030
	0.892	0.021	0.904	0.042	0.909	0.052
8	0.8892	0.0137	0.913	0.022	0.922	0
5	0.8910	0.0187	0.911	0.040	0.936	0.042
9	0.8896	0.0162	0.908	0.037	0.927	0.055
9	0.8900	0.018	0.9093	0.036	0.9274	0.055

ants [1,1,1] or [3,3,3]
el function [2/2] or [5/5]
Borel function [1,1,1] or [3,3,3]
3) - Numerical solution of the Schrödinger equation

1/*n*-expansion for a hydrogen atom in pa
for a ground state, $B = 0.5$ ($F_* = 0.2532$)

	$F = 0.16$		$F = 0.2$			
	E_I	$\Gamma/2$	E_I	$\Gamma/2$		
	0.732	0	0.738	0	0.752	0
6	0.742	0.032	0.730	0.029	0.724	0.034
7	0.747	0.022	0.756	0.037	0.762	0.032
3	0.7735	0	0.724	0.015	0.757	0
9	0.7478	0.0167	0.777	0.036	0.726	0.048
8	0.7478	0.0195	0.768	0.041	0.795	0.066
8	0.7491	0.019	0.7669	0.040	0.7873	0.065

ants [1,1,1] or [3,3,3]
el function [2/2] or [5/5]
Borel function [1,1,1] or [3,3,3]
3) - Numerical solution of the Schrödinger equation